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(4-Hydroxyphenoxy)acetic acid

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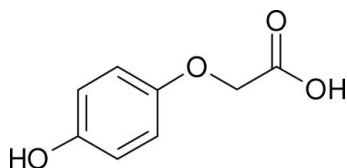
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Key indicators: single-crystal X-ray study; $T = 120$ K; mean $\sigma(\text{C}-\text{C}) = 0.001$ Å; R factor = 0.034; wR factor = 0.099; data-to-parameter ratio = 14.9.

The carboxyl groups in molecules of the title compound, $\text{C}_8\text{H}_8\text{O}_4$, are held together by $R_2^2(8)$ hydrogen bonding. The $-\text{OH}$ groups at position 4 on the aryl rings are linked by continuous hydrogen-bonded chains. The supramolecular structure is further defined by the formation of $R_6^6(44)$ rings.

Related literature

For related literature, see: Birner (1972); Cox & Hickey (2004); Hodgson & Asplund (1991); Kumar & Rao (1982).



Experimental

Crystal data

 $\text{C}_8\text{H}_8\text{O}_4$ $M_r = 168.14$ Monoclinic, $P2_1/c$ $a = 8.6550$ (2) Å $b = 5.2531$ (1) Å $c = 16.2401$ (4) Å $\beta = 98.983$ (1)° $V = 729.31$ (3) Å³ $Z = 4$ Mo $K\alpha$ radiation $\mu = 0.12$ mm⁻¹ $T = 120$ (2) K $0.6 \times 0.45 \times 0.38$ mm

Data collection

Bruker Nonius KappaCCD area-detector diffractometer

Absorption correction: multi-scan (SORTAV; Blessing, 1995)

 $T_{\min} = 0.844$, $T_{\max} = 0.954$ 9200 measured reflections
1666 independent reflections1518 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.034$ $wR(F^2) = 0.099$ $S = 1.07$

1666 reflections

112 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.35$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

 C_g is the centroid of the benzene ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O1}-\text{H1}\cdots\text{O1}^{\text{i}}$	0.84	2.07	2.8878 (6)	165
$\text{O3}-\text{H31}\cdots\text{O4}^{\text{ii}}$	0.84	1.78	2.6201 (10)	173
$\text{C5}-\text{H5}\cdots\text{O3}^{\text{iii}}$	0.95	2.49	3.4318 (12)	174
$\text{C2}-\text{H2}\cdots\text{Cg}^{\text{iv}}$	0.95	2.87	3.6190 (11)	136

Symmetry codes: (i) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 2, -y, -z$; (iii) $x - 1, y + 1, z$; (iv) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: DENZO (Otwinowski & Minor, 1997) and COLLECT (Nonius, 1998); cell refinement: DENZO and COLLECT; data reduction: DENZO and COLLECT; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: WinGX (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LW2013).

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supplementary materials

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(4-Hydroxyphenoxy)acetic acid

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Comment

4-hydroxyphenoxyacetic acid has been found in phenoxymethyl penicillin fermentations (Birner, 1972) and its molecular structure was determined to establish the hydrogen bonding patterns in the crystal. Apart from the methylene H atoms the molecule is essentially planar and the torsion angle with the greatest deviation from zero or $\pm 180^\circ$ is C7—O2—C1—C2 = 6.62 (14) $^\circ$. The molecular structure is shown in Figure 1, a partial packing diagram is given in Figure 2 and selected geometric parameters are listed in Table 1. Details of the hydrogen bonding are shown in Table 2.

As is often the case for monocarboxylic acids the carboxylate groups are linked by intermolecular hydrogen bonding to form a $R^2_2(8)$ ring across a centre of symmetry. Weaker hydrogen bonding is also present between the 4-hydroxy groups that link an infinite chain of molecules. Taken in combination the two hydrogen motifs also form $R^6_6(44)$ rings involving four molecules.

Other interactions involve a C5—H5 \cdots O3 contact and one C—H \cdots π bonding distance (Table 2) less than 3 Å (where Cg is the centre of the aryl ring).

Related structures include phenylacetic acid (Hodgson & Asplund, 1991), 4-methoxyphenoxyacetic acid (Kumar & Rao, 1982) and 2-methylphenoxyacetic acid (Cox & Hickey, 2004).

Experimental

The sample was purchased from Aldrich and was recrystallized from ethanol.

Refinement

All non-hydrogen atoms were refined by full-matrix least squares calculations with anisotropic displacement parameters. All the hydrogen atoms were allowed to ride on their attached atoms with isotropic displacement parameters 1.2 times those of the U_{eq} of their attached atoms. The constrained C—H distances were 0.95 Å (aromatic), 0.99 Å (methylene) and 0.84 Å (hydroxy).

Figures

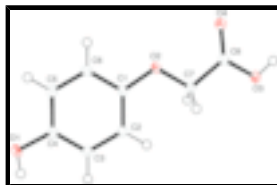


Fig. 1. View of (I) (50% probability displacement ellipsoids)



Fig. 2. A partial packing diagram of (I) showing $R^2_2(8)$ and $R^6_6(44)$ ring formations. Symbols used to denote symmetry positions of atom coordinates are: asterisk (*) = $2 - x, -y, -z$; dollar (\$) = $-x, 1/2 + y, 0.5 - z$ and hash (#) = $-x, -1/2 + y, 0.5 - z$.

(4-Hydroxyphenoxy)acetic acid

Crystal data

$C_8H_8O_4$	$F_{000} = 352$
$M_r = 168.14$	$D_x = 1.531 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
Hall symbol: -P 2ybc	$\lambda = 0.71073 \text{ \AA}$
$a = 8.6550 (2) \text{ \AA}$	Cell parameters from 1796 reflections
$b = 5.25310 (10) \text{ \AA}$	$\theta = 2.9\text{--}27.5^\circ$
$c = 16.2401 (4) \text{ \AA}$	$\mu = 0.12 \text{ mm}^{-1}$
$\beta = 98.9830 (10)^\circ$	$T = 120 (2) \text{ K}$
$V = 729.31 (3) \text{ \AA}^3$	Block, colourless
$Z = 4$	$0.6 \times 0.45 \times 0.38 \text{ mm}$

Data collection

Bruker Nonius KappaCCD area-detector diffractometer	1518 reflections with $I > 2\sigma(I)$
$T = 150(2) \text{ K}$	$R_{\text{int}} = 0.022$
φ and ω scans to fill Ewald sphere	$\theta_{\text{max}} = 27.5^\circ$
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	$\theta_{\text{min}} = 3.2^\circ$
$T_{\text{min}} = 0.844, T_{\text{max}} = 0.954$	$h = -10 \rightarrow 11$
9200 measured reflections	$k = -5 \rightarrow 6$
1666 independent reflections	$l = -20 \rightarrow 18$

Refinement

Refinement on F^2	H-atom parameters constrained
Least-squares matrix: full	$w = 1/[\sigma^2(F_o^2) + (0.0599P)^2 + 0.1898P]$
$R[F^2 > 2\sigma(F^2)] = 0.034$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.099$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.07$	$\Delta\rho_{\text{max}} = 0.35 \text{ e \AA}^{-3}$
1666 reflections	$\Delta\rho_{\text{min}} = -0.23 \text{ e \AA}^{-3}$
112 parameters	Extinction correction: none

Special details

Experimental. Please note cell_measurement_ fields are not relevant to area detector data, the entire data set is used to refine the cell, which is indexed from all observed reflections in a 10 degree phi range.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.03717 (8)	0.18406 (14)	0.22174 (5)	0.0183 (2)
H1	0.0231	0.0486	0.247	0.022*
O2	0.58461 (8)	0.16604 (13)	0.08499 (4)	0.0169 (2)
O3	0.92145 (8)	-0.20002 (14)	0.07019 (5)	0.0184 (2)
H31	0.9935	-0.1832	0.0414	0.022*
O4	0.84072 (8)	0.17346 (14)	0.01142 (4)	0.0188 (2)
C1	0.45139 (11)	0.15605 (18)	0.12213 (6)	0.0132 (2)
C2	0.42258 (11)	-0.03065 (19)	0.17835 (6)	0.0150 (2)
H2	0.4964	-0.163	0.1934	0.018*
C3	0.28426 (11)	-0.02249 (19)	0.21258 (6)	0.0152 (2)
H3	0.2645	-0.1483	0.2516	0.018*
C4	0.17624 (11)	0.16823 (18)	0.18973 (6)	0.0139 (2)
C5	0.20589 (11)	0.35600 (19)	0.13400 (6)	0.0158 (2)
H5	0.1319	0.4881	0.119	0.019*
C6	0.34355 (12)	0.35056 (18)	0.10027 (6)	0.0152 (2)
H6	0.3642	0.4793	0.0624	0.018*
C7	0.68607 (11)	-0.04464 (19)	0.10067 (6)	0.0153 (2)
H7A	0.6296	-0.2028	0.0813	0.018*
H7B	0.7228	-0.0601	0.1613	0.018*
C8	0.82384 (11)	-0.00948 (19)	0.05559 (6)	0.0145 (2)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0148 (4)	0.0192 (4)	0.0228 (4)	0.0023 (3)	0.0094 (3)	0.0016 (3)
O2	0.0130 (4)	0.0179 (4)	0.0213 (4)	0.0040 (3)	0.0076 (3)	0.0050 (3)
O3	0.0150 (4)	0.0204 (4)	0.0212 (4)	0.0059 (3)	0.0071 (3)	0.0051 (3)
O4	0.0168 (4)	0.0185 (4)	0.0223 (4)	0.0031 (3)	0.0069 (3)	0.0047 (3)
C1	0.0108 (5)	0.0154 (5)	0.0138 (4)	-0.0012 (3)	0.0029 (3)	-0.0022 (3)
C2	0.0134 (5)	0.0146 (5)	0.0171 (5)	0.0019 (3)	0.0023 (3)	0.0008 (3)
C3	0.0156 (5)	0.0151 (5)	0.0157 (4)	-0.0008 (3)	0.0044 (4)	0.0017 (3)
C4	0.0113 (4)	0.0163 (5)	0.0147 (4)	-0.0010 (3)	0.0034 (3)	-0.0030 (3)
C5	0.0146 (5)	0.0152 (5)	0.0177 (5)	0.0032 (3)	0.0028 (4)	0.0004 (3)
C6	0.0162 (5)	0.0141 (5)	0.0156 (4)	0.0004 (3)	0.0032 (4)	0.0015 (3)
C7	0.0127 (4)	0.0167 (5)	0.0173 (5)	0.0026 (4)	0.0042 (3)	0.0023 (4)
C8	0.0126 (4)	0.0173 (5)	0.0136 (4)	0.0004 (3)	0.0020 (3)	-0.0013 (3)

Geometric parameters (\AA , $^\circ$)

O1—C4	1.3861 (11)	C2—H2	0.95
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supplementary materials

O1—H1	0.84	C3—C4	1.3810 (14)
O2—C1	1.3832 (11)	C3—H3	0.95
O2—C7	1.4110 (12)	C4—C5	1.3894 (14)
O3—C8	1.3074 (12)	C5—C6	1.3871 (13)
O3—H31	0.84	C5—H5	0.95
O4—C8	1.2215 (12)	C6—H6	0.95
C1—C2	1.3887 (14)	C7—C8	1.5048 (13)
C1—C6	1.3920 (14)	C7—H7A	0.99
C2—C3	1.3971 (13)	C7—H7B	0.99
C4—O1—H1	109.5	C6—C5—C4	120.00 (9)
C1—O2—C7	115.42 (7)	C6—C5—H5	120
C8—O3—H31	109.5	C4—C5—H5	120
O2—C1—C2	124.12 (9)	C5—C6—C1	119.83 (9)
O2—C1—C6	115.61 (8)	C5—C6—H6	120.1
C2—C1—C6	120.27 (9)	C1—C6—H6	120.1
C1—C2—C3	119.53 (9)	O2—C7—C8	109.51 (8)
C1—C2—H2	120.2	O2—C7—H7A	109.8
C3—C2—H2	120.2	C8—C7—H7A	109.8
C4—C3—C2	120.10 (9)	O2—C7—H7B	109.8
C4—C3—H3	119.9	C8—C7—H7B	109.8
C2—C3—H3	119.9	H7A—C7—H7B	108.2
C3—C4—O1	122.26 (9)	O4—C8—O3	125.14 (9)
C3—C4—C5	120.26 (9)	O4—C8—C7	123.97 (9)
O1—C4—C5	117.47 (8)	O3—C8—C7	110.89 (8)
C7—O2—C1—C2	6.62 (14)	O1—C4—C5—C6	179.69 (8)
C7—O2—C1—C6	-173.88 (8)	C4—C5—C6—C1	0.27 (15)
O2—C1—C2—C3	179.73 (8)	O2—C1—C6—C5	179.67 (8)
C6—C1—C2—C3	0.25 (14)	C2—C1—C6—C5	-0.81 (15)
C1—C2—C3—C4	0.84 (14)	C1—O2—C7—C8	179.19 (7)
C2—C3—C4—O1	179.81 (8)	O2—C7—C8—O4	-1.16 (13)
C2—C3—C4—C5	-1.38 (15)	O2—C7—C8—O3	178.87 (8)
C3—C4—C5—C6	0.82 (15)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1—H1...O1 ⁱ	0.84	2.07	2.8878 (6)	165
O3—H31...O4 ⁱⁱ	0.84	1.78	2.6201 (10)	173
C5—H5...O3 ⁱⁱⁱ	0.95	2.49	3.4318 (12)	174
C2—H2...Cg ^{iv}	0.95	2.87	3.6190 (11)	136

Symmetry codes: (i) $-x, y-1/2, -z+1/2$; (ii) $-x+2, -y, -z$; (iii) $x-1, y+1, z$; (iv) $-x+1, y-1/2, -z+1/2$.

Fig. 1

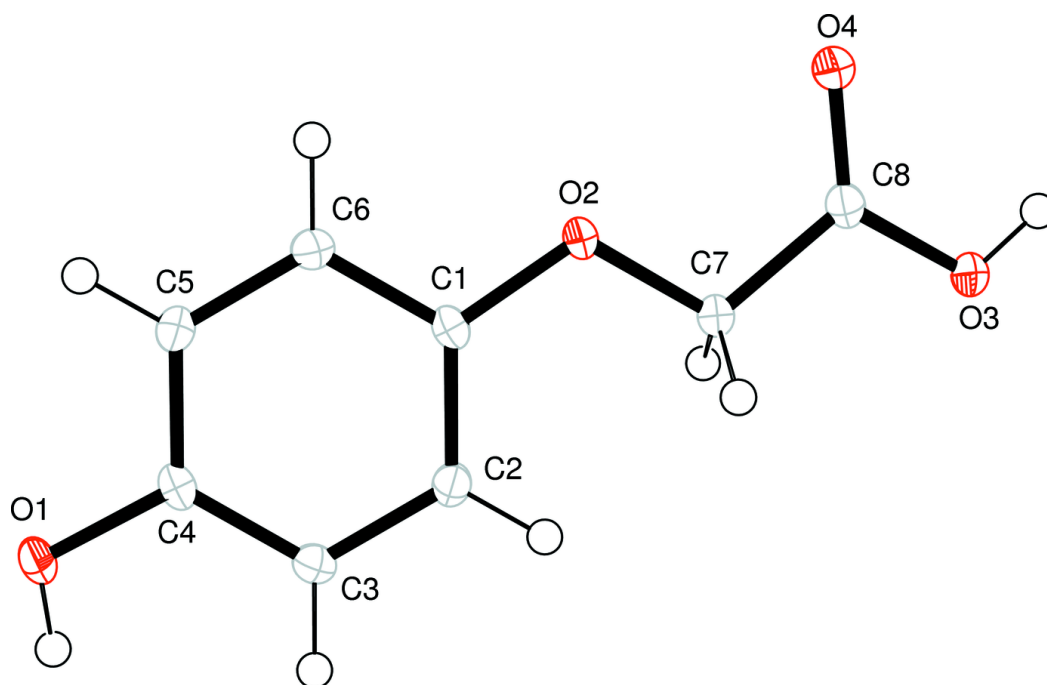


Fig. 2

